

STRUCTURES OF NAEMATOLIN AND NAEMATOLIN B,  
1S,9S-RING-FUSED CARYOPHYLLANE SESQUITERPENOIDS

Kozo DOI,\* Takao SHIBATA, Masataka NARA, Sei TSUBOYAMA,†  
Tosio SAKURAI,† and Kaoru TSUBOYAMA †

Department of Chemistry, Faculty of Science, Hirosaki  
University, Hirosaki-shi, Aomori 036

†The Institute of Physical and Chemical Research (Rikagaku  
Kenkyusho), Wako-shi, Saitama 351

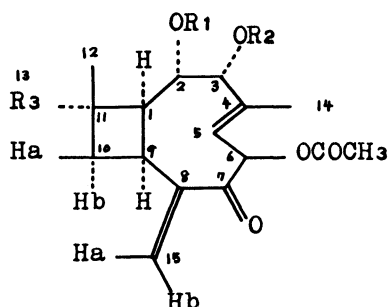
It is well known that naturally occurring caryophyllane  
sesquiterpenoids possess in general 1R,9S (trans)-ring juncture.  
However, it has been clarified that both of naematolin and naemato-  
lin B, microbial products, have 1S,9S (cis)-ring juncture in the  
caryophyllane skeleton.

Y. Ito *et al.*<sup>1)</sup> reported the isolation of naematolin from a liquid culture  
of mycelium of *Naematoloma fasciculare* (Nigakuritake in Japanese), although its  
structure was not deduced at that time. The authors reported, in the course of  
investigating metabolites of this fungus,<sup>2)</sup> the isolation and structural elucidation  
of two compounds. The first one (naematolin, 1), C<sub>17</sub>H<sub>24</sub>O<sub>5</sub>,<sup>3)</sup> mp 144-145 °C,  
[α]<sub>D</sub><sup>22</sup> -330°, afforded in a usual way an acetate (2), C<sub>21</sub>H<sub>28</sub>O<sub>7</sub>, mp 134-135 °C,  
[α]<sub>D</sub><sup>22</sup> -260°, which was identical with naematolin acetate<sup>1,4)</sup> by means of their  
mixed mp test and of the comparison of each <sup>1</sup>H-NOE. The second one, that was  
designated as naematolin B (5, mp 122-123 °C, and [α]<sub>D</sub><sup>23</sup> -352°), has a molecular  
formula of C<sub>17</sub>H<sub>24</sub>O<sub>6</sub>, indicative of a hydroxynaematolin. We wish to describe

herein the structures of naematolin  
(1) and naematolin B (5).

In order to elucidate the stereo-  
chemistry of 1, the following deriva-  
tives were prepared: naematolin di-  
benzoate (3), C<sub>31</sub>H<sub>32</sub>O<sub>7</sub>, mp 146-147 °C,  
and naematolin mono *p*-bromobenzoate  
(4), C<sub>24</sub>H<sub>27</sub>O<sub>6</sub>Br, mp 122-123 °C.

<sup>1</sup>H- and <sup>13</sup>C-NMR data of 2, 4, and  
5 are cited in Table 1, wherein the  
assignments of each signal have been  
carried out using <sup>1</sup>H spin-spin  
decoupling, 2D <sup>1</sup>H-<sup>1</sup>H COSY, INEPT, and  
2D <sup>1</sup>H-<sup>13</sup>C COSY.<sup>5)</sup>



- 1: R<sub>1</sub>=R<sub>2</sub>=H, R<sub>3</sub>=CH<sub>3</sub>  
2: R<sub>1</sub>=R<sub>2</sub>=CH<sub>3</sub>CO, R<sub>3</sub>=CH<sub>3</sub>  
3: R<sub>1</sub>=R<sub>2</sub>=C<sub>6</sub>H<sub>5</sub>CO, R<sub>3</sub>=CH<sub>3</sub>  
4: R<sub>1</sub>=H, R<sub>2</sub>=*p*-BrC<sub>6</sub>H<sub>4</sub>CO, R<sub>3</sub>=CH<sub>3</sub>  
5: R<sub>1</sub>=R<sub>2</sub>=H, R<sub>3</sub>=CH<sub>2</sub>OH

Table 1.  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR of 2, 4, and 5

No. of C	<u>2</u>		<u>4</u>	<u>5</u>	
	$^{13}\text{C}$	$^1\text{H}$ (J)	$^1\text{H}$ (J)	$^{13}\text{C}$	$^1\text{H}$ (J)
1	44.9 d	3.09 t(11)	2.90 t(11)	45.1 d	3.12 t(11)
2	70.8 d	4.86 dd(12,2)	3.89 dd(12,2)	70.6 d	3.55 dd(11,1.5)
3	75.0 d	5.23 d(2)	5.23 d(2)	78.1 d	4.06 d(1.5)
4	143.3 s			148.6 s	
5	117.8 d	6.10 d(9)	6.08 d(9)	117.1 d	6.32 d(9)
6	71.0 d	5.76 d(9)	5.76 d(9)	72.8 d	5.82 d(9)
7	195.2 s			196.8 s	
8	148.4 s			151.0 s	
9	35.1 d	3.18 td(10,5)	3.19 td(10,5)	36.6 d	3.31 td(11,5)
10a		1.85 dd(12,5)	1.88 dd(12,5)		1.69 dd(13,5)
10	40.9 t			36.8 t	
10b		2.23 dd(12,10)	2.23 dd(12,10)		2.24 dd(13,10)
11	34.6 s			40.5 s	
12	23.3 q	1.05 s	1.27 s	19.7 q	1.30 s
13	33.0 q	1.22 s	1.32 s	73.3 t	3.42 AB(2H,9.8)
14	18.2 q	1.67 br.s	1.69 br.s	19.3 q	1.52 d(1.5)
15a		5.80 s	5.84 s		5.83 s
15	127.3 t			126.6 t	
15b		6.56 s	6.55 s		6.42 s
<u>CH<sub>3</sub>CO</u>	21.1 q	2.01 s	2.16 s	21.1 q	2.11 s
<u>CH<sub>3</sub>CO</u>	20.6 q	2.17 s			
<u>CH<sub>3</sub>CO</u>	20.9 q	2.19 s			
<u>CH<sub>3</sub>CO</u>	169.4 s			170.5 s	
<u>CH<sub>3</sub>CO</u>	169.8 s				
<u>CH<sub>3</sub>CO</u>	170.0 s				
OH			1.60 br.s (1H)		2.84 d(3H)
others			7.66 d(2H,8.5)		
			7.95 d(2H,8.5)		

The spectra were taken on a JEOL JNM-GX400 in the each solvent(2 and 4: chloroform-d; 5: acetone-d<sub>6</sub>).

It is evident from the above Table 1 that the geometry of 4-endocyclic double bond of 1 is trans based on the chemical shift of 14-methyl in  $^{13}\text{C}$ -NMR.<sup>6)</sup> In addition, *p*-bromobenzoyloxyl group in 4 and the third hydroxyl one in 5 must be attached to C(3) and C(13), respectively.

The circular dichroism spectrum<sup>7)</sup> of 3 showed that the first Cotton effect was negative ( $\Delta\epsilon_{241}$ , -33.9) and the second one was positive ( $\Delta\epsilon_{220}$ , +15.2), indicating that 3 has negative chirality in the light of the dibenzoate exciton chirality rule.<sup>8)</sup> From this fact, 1 possesses clearly the 2S,3R-glycol system.

In order to determine the stereochemistry of the additional chiral centers in 1 except for C(2) and C(3), NOE difference spectra of 2 were measured,<sup>9)</sup> and the results were summarized in Table 2. On irradiation at 12-H, NOEs were observed not only at 2-H but at 15-Ha, suggesting that both of C(1)-C(2) and

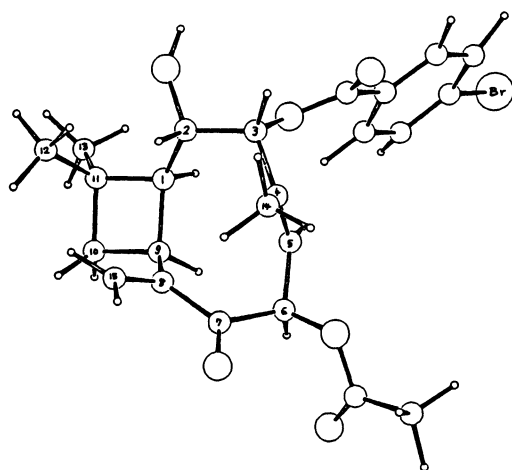
Table 2. NOE of 2 and 5

Compound	Irradiated	Observed	NOE/%
<u>2</u>	12-H	2-H	10
	12-H	15-Ha	10
	13-H	1-H	25
	14-H	2-H	15
	14-H	3-H	15
	9-H	6-H	10
	9-H	5-H	10
	2-H	3-H	13
	15-Ha	15-Hb	32
	5-H	6-H	11
	5-H	9-H	12
	5-H	1-H	22
	<u>5</u>	12-H	15-Ha
12-H		2-H	5
1-H		9-H	15
1-H		13-H	21
9-H		1-H	3
9-H		6-H	6
6-H		5-H	6
5-H	6-H	7	

C(8)-C(9) bondings are in a cis configuration, and accordingly the juncture between the four- and nine-membered rings is also cis. Furthermore, NOEs by the irradiations of 5- and 9-H verify that the four protons at 1-, 5-, 6-, and 9-positions adopt the same configuration. Therefore, the structure of naematolin must be depicted as 1, involving 1S,2S,3R,6R, and 9S configurations.

This proposal has been well confirmed by an X-ray analysis of 4. A perspective drawing is given in Fig. 1, in which the ring juncture is cis, the geometry of the endocyclic double bond is trans, and the configuration is 1S,2S,3R,6R, and 9S, respectively. This finding agrees well with the absolute configuration deduced from the above spectral data.

Crystal data were as follows: C<sub>24</sub>H<sub>27</sub>O<sub>6</sub>Br (MW=491.36), orthorhombic, space group P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>, a=10.650(2), b=20.650(6), c=10.526(2) Å<sup>3</sup>, Z=4, D<sub>c</sub>=1.410 Mg m<sup>-3</sup>, μ=1.791 mm<sup>-1</sup>. A total of 1666 reflections was recorded on a RIGAKU four-circle diffractometer with graphite-monochromatized MoKα (λ=0.71073 Å) radiation. The structure was solved by the direct method and the final R value was 0.084. The absolute configuration was determined through the anomalous dispersion effect of the bromine atom.

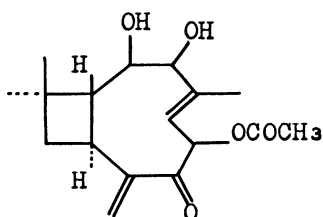
Fig. 1. A perspective drawing of 4.

The structure of naematolin B, which was deduced as 13-hydroxy-

naematolin from Table 1, agrees well with 5 from the NOE data cited in Table 2. An example of 12-hydroxycaryophyllane derivative has been reported as a metabolite of a mushroom.<sup>10)</sup>

Naematolin(1) and naematolin B(5) both demonstrate the first case of 1-epi-caryophyllane skeleton in nature, although 9-epi-caryophyllane derivatives are known as plant constituents.<sup>11)</sup>

Backens et al.<sup>12)</sup> have proposed an alternate structure (6) for naematolin without any direct comparison with Ito's compound.<sup>1)</sup> Presumably, the Backens' compound is 1-epi-naematolin.



6

#### References

- 1) Y.Ito, H.Kurita, T.Yamaguchi, M.Sato, and T.Okuda, Chem.Pharm.Bull., 15, 2009(1967); Y.Ito, H.Kurita, and M.Sato, Japan Patent, No.16795(1970).
- 2) K.Do, T.Shibata, and M.Nara, Abstract Papers Chem.Soc. Meeting in Tohoku District, 1984, pp.57, 73; 1985, pp.39, 40.
- 3) All molecular formulas were obtained by HRMS.
- 4) We wish to thank Dr.H.Kurita of Microbial Chem. Res. Lab., Tanabe Seiyaku Co.,Ltd. for his kindly gift of naematolin acetate.
- 5) 2D COSYS were measured on a JEOL JNM-GX 270. We thank Messr. K.Hioka, N.Fujii, and O.Kamo of NMR Application Lab., JEOL Ltd. for the measurements.
- 6) Y.Kashman and A.Groweiss, J.Org.Chem., 45, 3814(1980).
- 7) It was recorded on a JASCO J-20A spectropolarimeter in MeOH.
- 8) N.Harada and K.Nakanishi, "Circular Dichroic Spectroscopy—Exciton Coupling in Organic Stereochemistry" (in Japanese), Tokyo Kagaku Dojin, Tokyo, 1982, p.31.
- 9) We thank Dr.J.Uzawa and Mrs.T.Chijimatsu of this Institute for the measurements.
- 10) W.M.Daniewski, P.A.Grieco, J.C.Huffman, A.Rymkiewicz, and A.Wawrzum, Phytochem., 20, 2733(1981).
- 11) F.Bohlmann and C.Zdero, Phytochem., 17, 1135(1978); F.Bohlmann and J.Ziesche, *ibid.*, 20, 469(1981).
- 12) S.Backens, B.Steffan, W.Steglich, L.Zechlin, and T.Anke, Liebigs Ann.Chem., 1984, 1332.

( Received January 7, 1986 )